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Single Replacement Reaction Worksheet Pdf

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1. Balance the following:

a) 
$$CH_4 + 2O_2 \rightarrow CO_2 + 2H_2O$$

e) 
$$2CO_2 \rightarrow 2CO + O_2$$

2. Lithium oxide + Water → Lithium hydroxide

Hydrogen peroxide → Oxygen + Water

4. Iron (III) + Hydrogen chloride → Hydrogen + Iron (III) chloride

5. Sodium sulphate + Barium nitrate -> Sodium nitrate + Barium sulphate

6. Hydrogen nitrate + Barium hydroxide → Water + Barium nitrate

a) Skeleton: 
$$\frac{H \times N0_3 + Ba(0H)_2 \rightarrow H0H + Ba(N0_3)_2}{2 + N0_3 + Ba(0H)_2 \rightarrow 2 + H0H + Ba(N0_3)_2}$$
b) Balanced: 
$$\frac{2 + N0_3 + Ba(0H)_2 \rightarrow 2 + H0H + Ba(N0_3)_2}{2 + H0H + Ba(N0_3)_2}$$

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### single replacement reaction worksheet key

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