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Pulsed Nuclear Magnetic Resonance

Experiment NMR

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References

- 3rd ed. 1990)
- E. Fukushima, S. B. W. Roeder, Experimental Pulse NMR: A nuts and bolts approach, (Perseus Books, 1981)
- Physics, (Addison Wesley, 1974).
- Physics,

Introduction

To observe nuclear magnetic resonance, the sample nuclei are first aligned in a strong magnetic field. In this experiment, you will learn the techniques used in a pulsed nuclear magnetic resonance apparatus (a) to perturb the nuclei out of alignment with the field and (b) to measure the small return signal as the misaligned nuclei precess in the field.

Because the return signal carries informabetter than 1 ppm, to four figures $\gamma=2.675\times$ tion about the nuclear environment, the tech- 10^8 rad/(see tesla). nique has become widely used for material analysis. Initially, you will study the behavior of hydrogen nuclei in glycerin because the signal is easy to find and interpret. Samples involving other nuclei and other nuclear environments can also be investigated.

Theory

C. P. Schlicter, Principles of Magnetic Res- Recall that the hydrogen nucleus consists of a onance, (Springer, Berlin, 2nd ed. 1978, single proton and no neutrons. The precession of a bare proton in a magnetic field is a simple consequence of the proton's intrinsic angular momentum and associated magnetic dipole moment. A classical analog would be a gyroscope having a bar magnet along its rotational M. Sargent, M.O. Scully, W. Lamb, Laser axis. Having a magnetic moment, the proton experiences a torque in a static magnetic field. Having angular momentum, it responds to the A. C. Melissinos, Experiments in Modern torque by precessing about the field direction. This behavior is called Larmor precession.

The model of a proton as a spinning positive charge predicts a proton magnetic dipole moment μ that is aligned with and proportional to its spin angular momentum ${f s}$

$$\mu = \gamma s$$
 (1)

where γ , called the gyromagnetic ratio, would depend on how the mass and charge is distributed within the proton. Determined by various magnetic resonance experiments to

Note that different nuclei will have different spin angular momentum, different magnetic moments, and different gyromagnetic ratios. The gyromagnetic ratio given above is only for the hydrogen nucleus. In NMR, the word proton generally only refers to the nucleus of the

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