

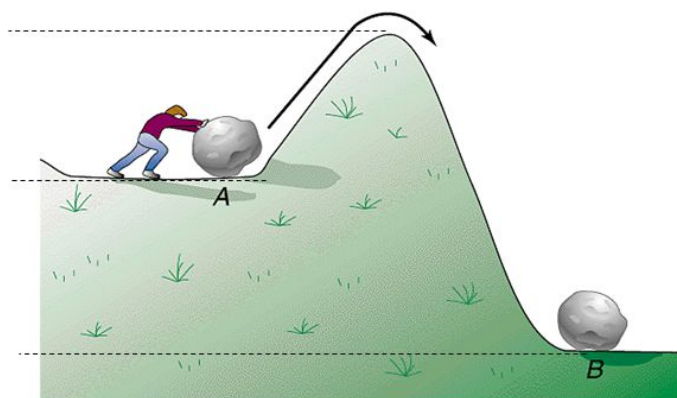


Activation Energy

Definition

Activation Energy : The energy required to overcome the reaction barrier. Usually given a symbol E_a or ΔG^\ddagger

The Activation Energy (E_a) determines how fast a reaction occurs, the *higher* Activation barrier, **the slower the reaction rate**. **The lower the Activation barrier, the faster the reaction**



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